

Surprises in patchy colloids

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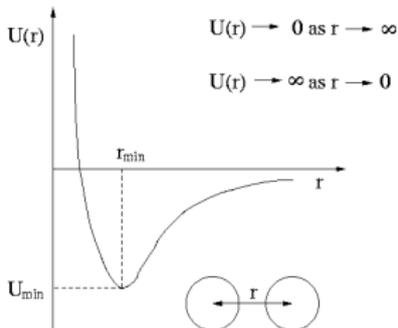
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Phases and phase transitions

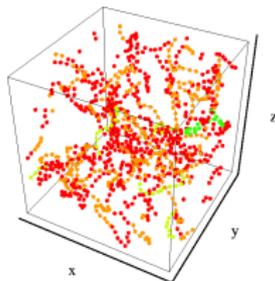
- In general, interactions between atoms, molecules or ions comprise a **short-range repulsion** and a **longer-ranged attraction**.



- In the **Van der Waals picture**, the interplay of **interaction energy** and **entropy** – expressed as **free energy** – determines which **phases** are realised.
 - At **low temperatures/high densities** energy wins and we have a **condensed** phase – a liquid or a solid.
 - At **high temperatures/low densities** entropy wins and we have a **dilute** phase – a gas.

Motivation

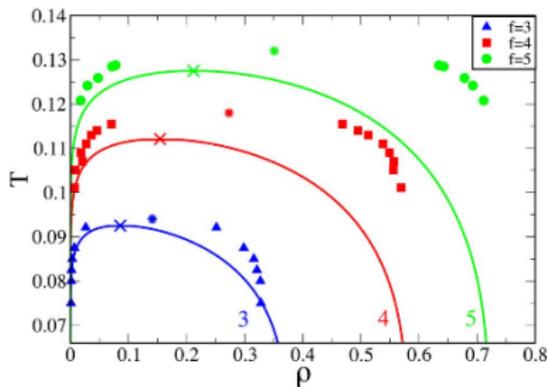
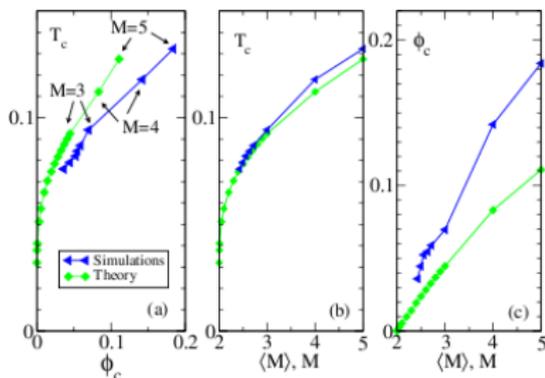
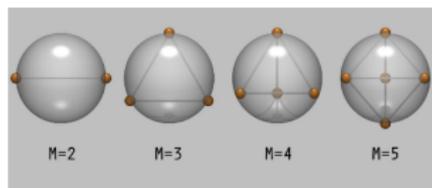
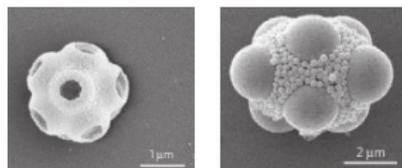
- Most molecular species have **permanent dipoles**.
- **Criticality of strongly dipolar fluids** is still **unsolved problem** (image by J.-J. Weis).



- A related, more general issue is **interplay between condensation and association**.
- We want to study a model that **retains the essential symmetry** of dipolar forces leading to association, but **leaves out apparently inessential features** (long range and complex angular dependence).

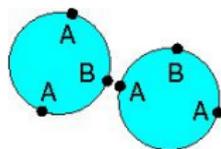
Patchy colloids

- Patchy colloids are custom-fabricated matter that exhibits both self-assembly and the usual phase transitions (condensation, freezing, etc) (images by Y. S. Cho *et al.*).
- Sciortino *et al.* [Phys. Rev. Lett. **97**, 168301 (2006); J. Chem. Phys. **128**, 144504 (2008)] simulated patchy particles with M identical sites: $\rho_c \rightarrow 0$ and $T_c \rightarrow 0$ as $M \rightarrow 2$.



Theory and model

- **Hard spheres** of diameter σ and volume v_s , each decorated with $2 + m_B$ sticky spots: two A's and m_B B's. AA, BB or AB bonds may form.



- **Bonding free energy** from Wertheim's theory:

$$\beta f_b \equiv \frac{\beta F_b}{N} = 2 \ln X_A + m_B \ln X_B - X_A - \frac{m_B X_B}{2} + \frac{2 + m_B}{2}$$

X_i is the probability of having a sticky spot of type i not bonded.

- **Law of mass action** yields:

$$X_A + 2\eta\Delta_{AA}X_A^2 + m_B\eta\Delta_{AB}X_AX_B = 1$$

$$X_B + m_B\eta\Delta_{BB}X_B^2 + 2\eta\Delta_{AB}X_AX_B = 1,$$

where $\eta \equiv (N/V)v_s$ is the packing fraction, and

$$\Delta_{ij} = \frac{1}{v_s} \int_{v_{ij}} g_{ref}(\mathbf{r}) [\exp(-\beta\epsilon_{ij}) - 1] d\mathbf{r}$$

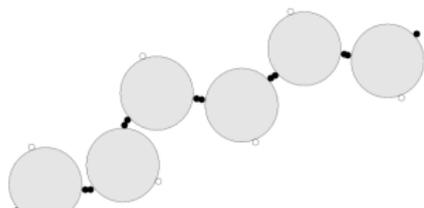
- **Free energy per particle** is a function of (η, T) only:

$$\beta f = \beta f_{HS} + \beta f_b$$

Ground states (without loops)

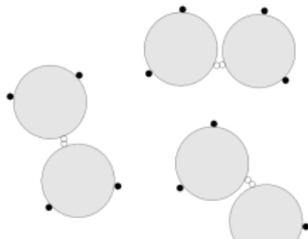
Linear chains

$$(\epsilon_{AB} = \epsilon_{BB} = 0, \epsilon_{AA} \neq 0)$$



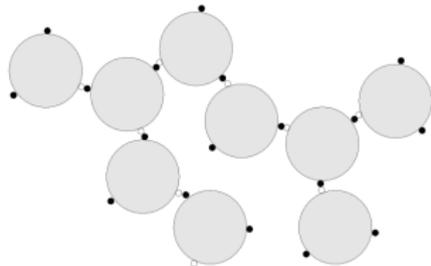
Dimers ($m_B = 1$ only!)

$$\epsilon_{AA} = \epsilon_{AB} = 0, \epsilon_{BB} \neq 0$$



Hyperbranched polymers

$$(\epsilon_{AA} = \epsilon_{BB} = 0, \epsilon_{AB} \neq 0)$$



Elementary excitations

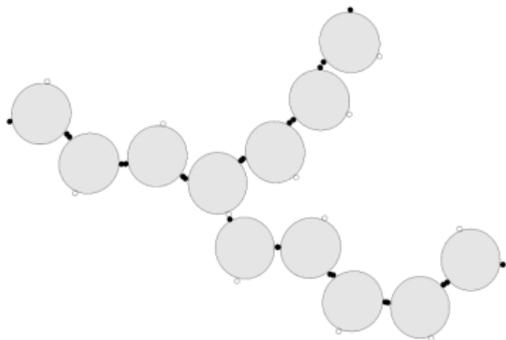
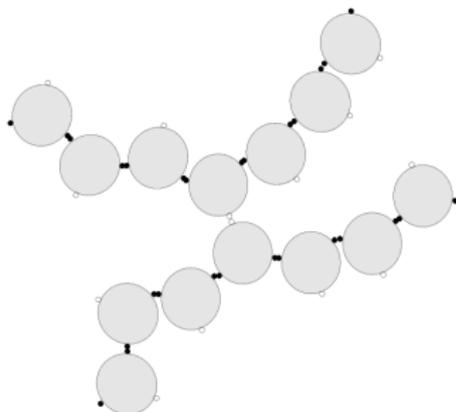
X-junction is always favourable:

$$\epsilon_j = -\epsilon_{BB} < 0$$

Y-junction is favourable only if

$$\epsilon_j = -\epsilon_{AB} + \epsilon_{AA}/2 < 0$$

$$\Leftrightarrow \epsilon_{AB}/\epsilon_{AA} > 1/2$$



Asymptotic behaviour

- X-junction driven criticality: $\epsilon_{AB} = 0$, $\epsilon_{BB} \rightarrow 0$

$$T_c = \frac{\epsilon_{BB}}{\ln b},$$
$$\eta_c = \left[\frac{9v_b}{8v_s(B_3 + 6B_2^2)^2} \right]^{\frac{1}{5}} \exp \left[-\frac{\ln b}{5(\epsilon_{BB}/\epsilon_{AA})} \right].$$

A critical point is always present, with lower and lower critical density and temperature.

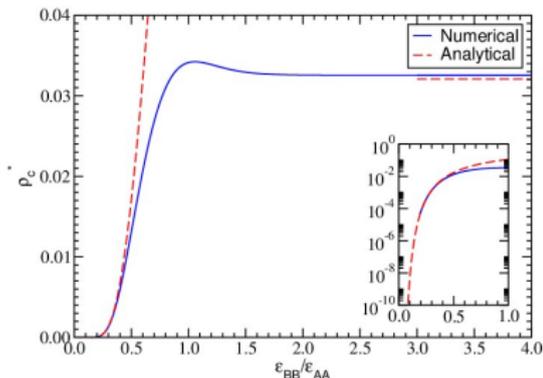
- Y-junction driven criticality: $\epsilon_{BB} = 0$, $\epsilon_{AB} \rightarrow 0$

$$T_c = \frac{\epsilon_{AB} - \frac{1}{3}\epsilon_{AA}}{b},$$
$$\eta_c = \frac{v_b}{v_s} \exp \left(-\frac{b\epsilon_{AB}}{3\epsilon_{AB} - \epsilon_{AA}} \right).$$

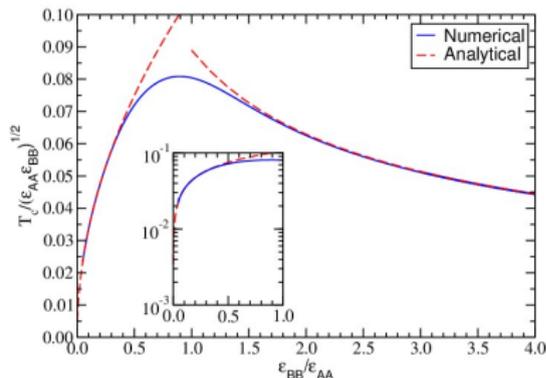
On decreasing ϵ_{AB} a critical point of vanishingly small density and temperature is obtained only up to $\epsilon_{AB}/\epsilon_{AA} = \frac{1}{3}$.

X and Y criticality

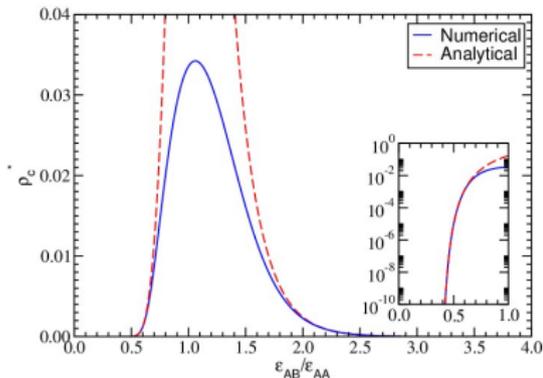
(a)



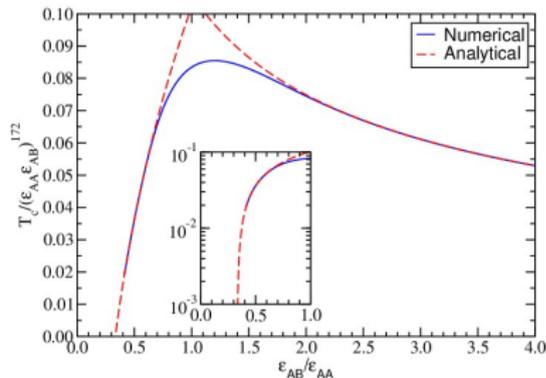
(b)



(a)



(b)

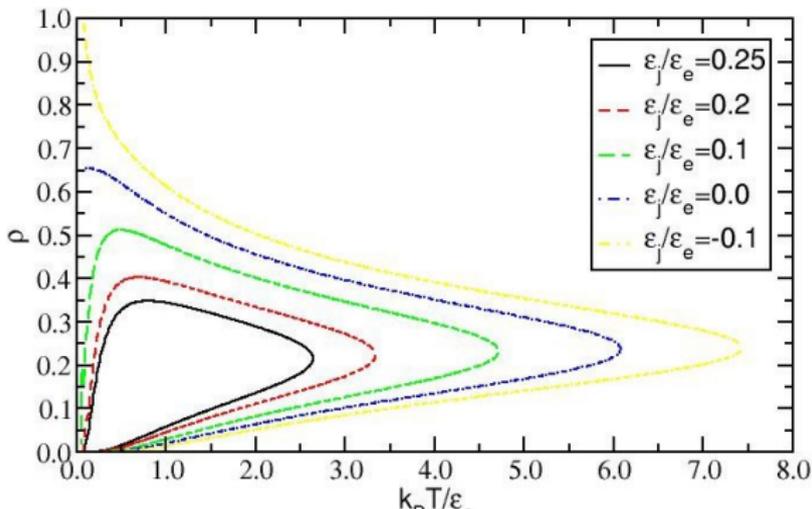


Summary of results so far

- If $\epsilon_{AA} = 0$ there is **no critical point**.
- If $\epsilon_{AA} \neq 0$:
 - If $\epsilon_{AB} = 0$, the critical point exists all the way to $\epsilon_{BB}/\epsilon_{AA} = 0$. **This corresponds to X-junction condensation.**
 - If $\epsilon_{BB} = 0$, there is no critical point for $\epsilon_{AB}/\epsilon_{AA} < 1/3$. **This corresponds to Y-junction condensation.**
- By changing the ratio of interaction strengths, we are able to engineer very low density liquid phases: **'empty liquids'**.
- Likewise, different **cluster structures** may result, which may or may not lead to **percolation**.
- But wait... What happens if $1/3 < \epsilon_{AB}/\epsilon_{AA} < 1/2$?

Diversion: Tlusty and Safran's theory

- At low T (large μ), dipolar fluid consists mostly of **long chains**.
- Treated as a **perturbation** of a ground state of **infinitely long chains**.
- Perturbation consists of two types of **thermally-excited defects**:
chain ends , of energy ϵ_e and **Y-junctions**  of energy ϵ_j .
- There is a **critical point** if $\epsilon_j/\epsilon_e < 3$. Coexistence is between a **lower-density** phase rich in ends and a **higher-density** phase rich in junctions. Phase diagram **pinches (is re-entrant)** at low T .



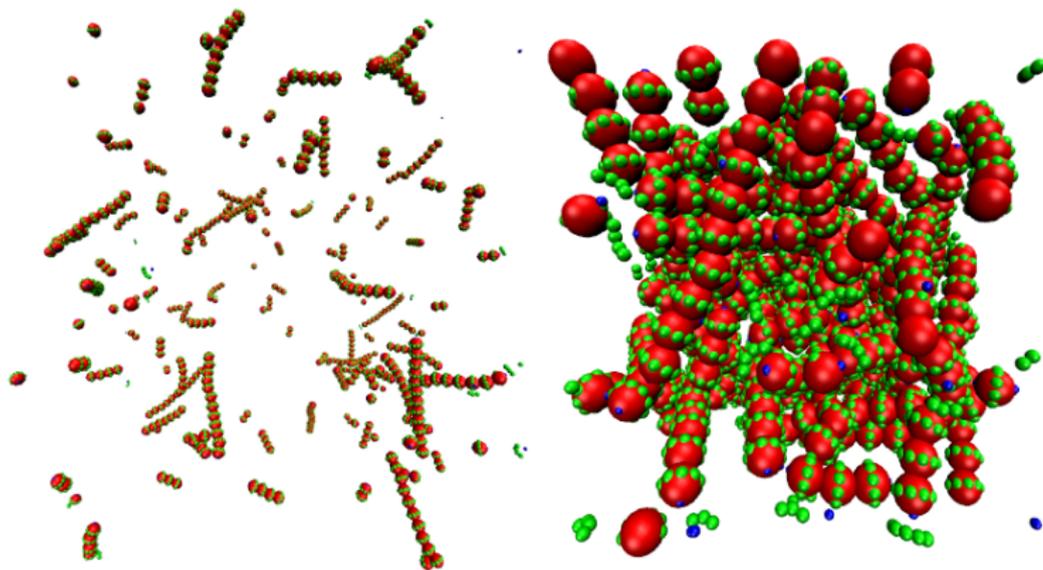
Issues to be addressed ($\epsilon_{BB} = 0$)

- 1 Can we also find **pinching** in our theory?
 - TS theory **coincides** with ours in limit $\epsilon_{AB}/\epsilon_{AA} \ll 1$. It affords **greater insight** into our own theory, **BUT**:
 - TS theory is **lattice-based** \Rightarrow **not-so-good entropy**.
 - ϵ_e and ϵ_j **are not related to interparticle potentials**.
 - **BUT** we need to be able to approach **limit where critical point disappears**, where vapour densities are **extremely low**.

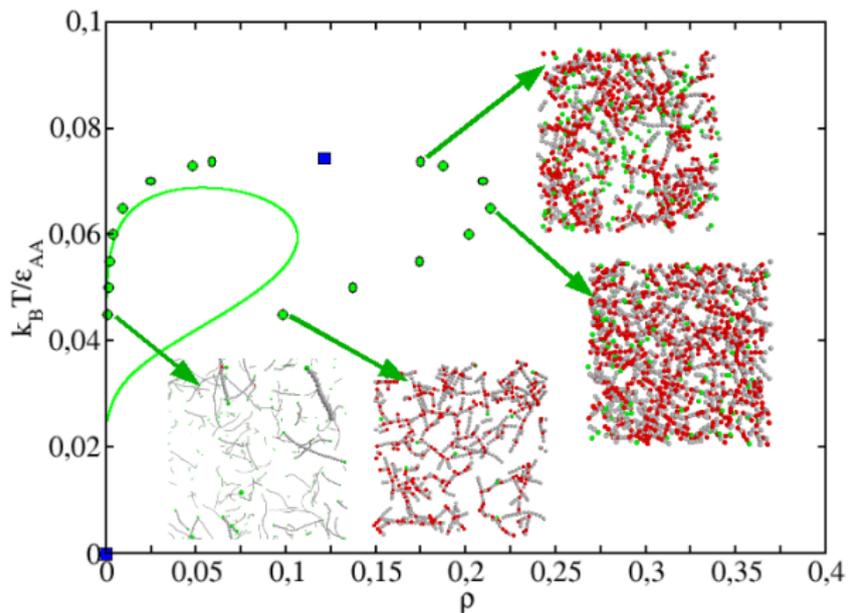
Solution: **Artificially increase** AB -bond volume by choosing $m_B = 9$: not one larger patch, but **many small ones**.
- 2 How do we go from 'pinched' to 'normal' phase behaviour?

Solution: Switch on ϵ_{BB} at fixed $1/3 < \epsilon_{AB}/\epsilon_{AA} < 1/2$.

Snapshot of vapour and liquid phases

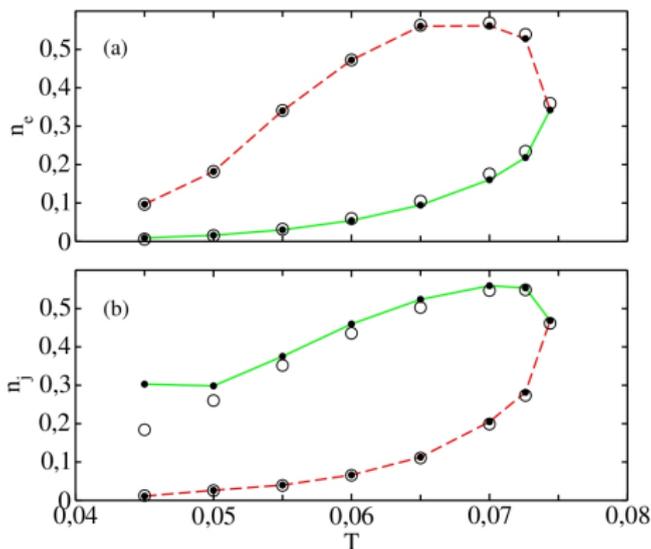


Phase diagram



- T_c^* well predicted, ρ_c^* less so.
- Vapour phase **rich in ends**, liquid phase **rich in junctions**.

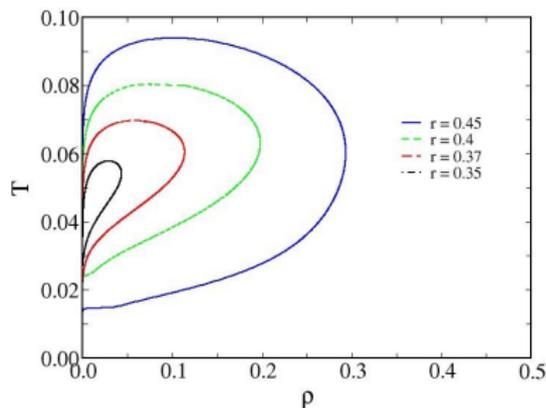
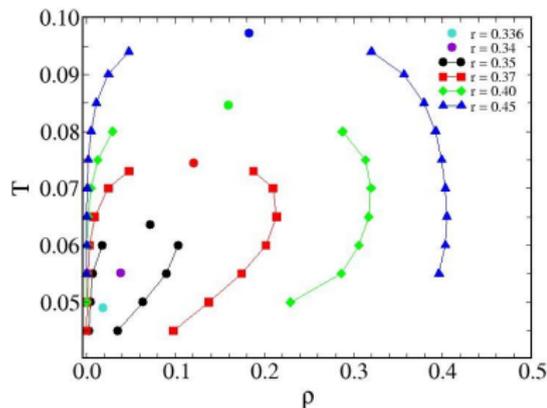
Fractions of chain ends and junctions along coexistence line



- Excellent agreement between theory and simulation (using simulation input).
- Vapour phase rich in ends, liquid phase rich in junctions.

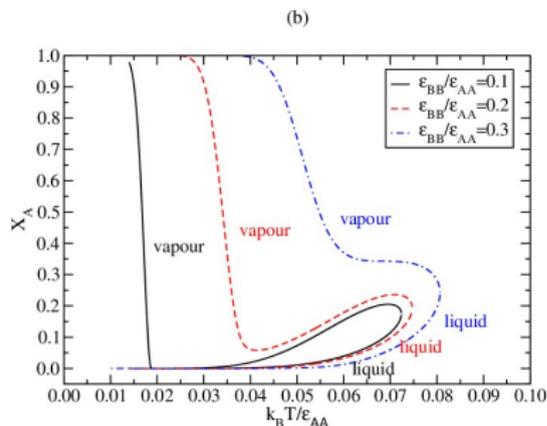
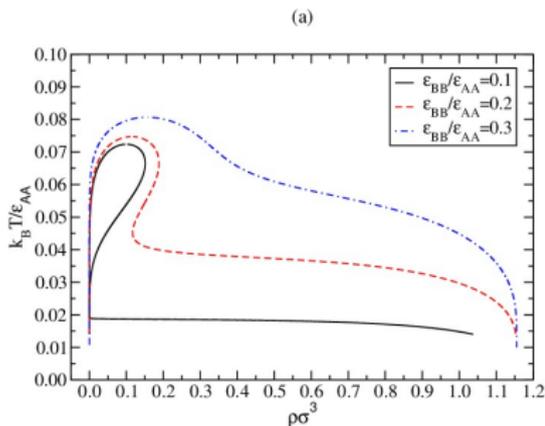
Results for $2A + 9B$ model, $\epsilon_{BB} = 0$, variable ϵ_{AB}

Phase diagram



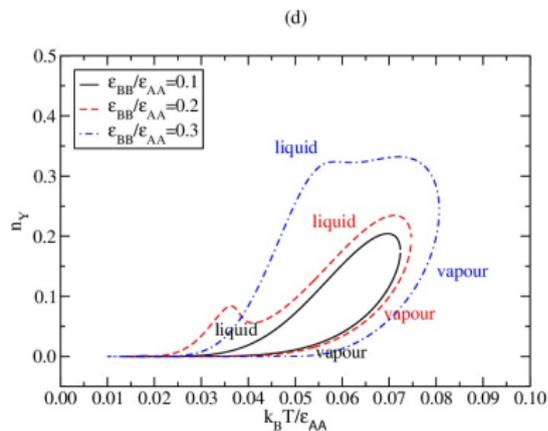
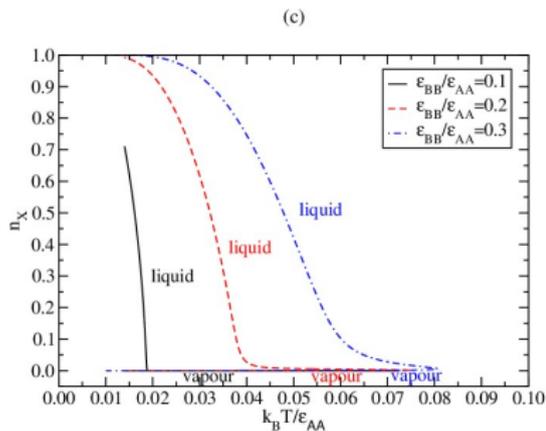
- For all $\epsilon_{AB}/\epsilon_{AA}$ a **clear pinching** is observed, which becomes more pronounced as $\epsilon_{AB}/\epsilon_{AA} \rightarrow 1/3$.
- On decreasing $\epsilon_{AB}/\epsilon_{AA}$, **both T_c and ρ_c decrease**.
- Theory **correctly predicts the temperature range of condensation**, but **significantly underestimates ρ_c and the density of the liquid-branch of the binodal**.

Phase diagram and fraction of unbonded A's



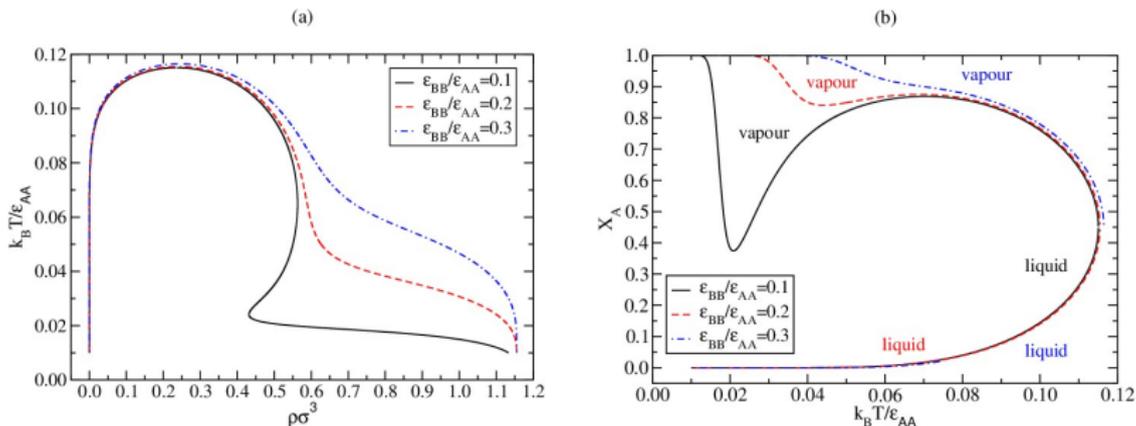
- Phase diagram gradually **un-pinches** as $\epsilon_{BB} / \epsilon_{AA}$ increases.
- In the re-entrant region for small $\epsilon_{BB} / \epsilon_{AA}$, both phases consists almost exclusively of **extremely long chains** ($X_A \approx 0$).

Fractions of X- and Y-junctions



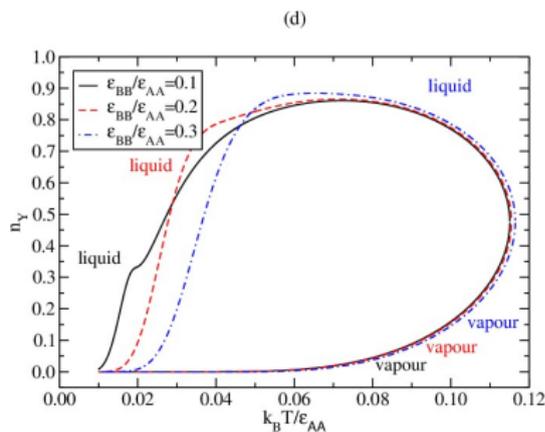
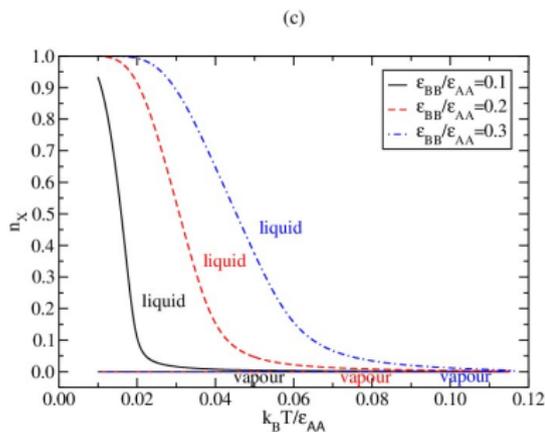
- Close to T_c , coexistence is between a liquid of long chains and rich in Y-junctions, and a vapour of shorter chains with fewer Y-junctions, both with practically no X-junctions
- As $T \rightarrow 0$, the liquid has many X-junctions and no Y-junctions, the vapour is an ideal gas of monomers.
- At intermediate T , coexistence is between a gas of short chains and a liquid of very long chains with X- and, in some cases, Y-junctions.

Phase diagram and fraction of unbonded A's



- Phase diagram gradually **un-pinches** as $\epsilon_{BB} / \epsilon_{AA}$ increases.
- In the re-entrant region for small $\epsilon_{BB} / \epsilon_{AA}$, both phases consists almost exclusively of **extremely long chains** ($X_A \approx 0$).

Fractions of X- and Y-junctions



- **Close to T_c** , coexistence is between a liquid of long chains and rich in Y-junctions, and a vapour of shorter chains with fewer Y-junctions, both with practically no X-junctions
- **As $T \rightarrow 0$** , the liquid has many X-junctions and no Y-junctions, the vapour is an ideal gas of monomers.
- **At intermediate T** , coexistence is between a gas of short chains and a liquid of very long chains with X- and, in some cases, Y-junctions.

Summary and conclusions

- We have applied **Wertheim's theory of association** to patchy colloids with two A sites and m_B B sites.
- For $\epsilon_{BB} \rightarrow 0$ or $\epsilon_{AB} \rightarrow 0$, **long AA chains form** with either AB or BB branches. These are relevant to strong-dipolar-fluid criticality.
- When $\epsilon_{BB} = \epsilon_{AB} = 0$ we recover the non-trivial limit of two A 's:
 - If $\epsilon_{AB} = 0$, the critical point exists all the way to $\epsilon_{BB}/\epsilon_{AA} = 0$. **This corresponds to X-junction condensation.**
 - If $\epsilon_{BB} = 0$, there is no critical point for $\epsilon_{AB}/\epsilon_{AA} < 1/3$. **This corresponds to Y-junction condensation.**
- We have been able to reproduce the pinched phase diagram of Tlusty-Safran theory using our **patchy particles** and **Wertheim's theory of association**.
- This re-entrance can be understood as a **temperature controlled effective valence**: the number of bonded sites per particle goes down with decreasing T .
- The structure of the coexisting phases can be understood simply by noting that **Y-junctions are favoured at high temperatures, whereas X-junctions dominate at low temperatures.**

BUT this version of the theory ignores completely:

- higher densities;
- rings;
- patch positions on the hard core \Rightarrow orientational correlations.

- J. M. Tavares, P. I. C. Teixeira and M. M. Telo da Gama, *Molec. Phys.* **107**, 453–466 (2009).
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- J. Russo, J. M. Tavares, P. I. C. Teixeira, M. M. Telo da Gama and F. Sciortino, *Phys. Rev. Lett.* **106**, 085703 (2011).
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- J. M. Tavares and P. I. C. Teixeira, *J. Phys.: Condens. Matter* **24**, 284108 (2012).

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